Lewis structures (Level 2) exam tips

• Draw clear diagrams, not undecipherable scribbles.

• Use sensibly sized dots, not dots that look like a speck of dust.

• In past NCEA Exams both dots and crosses have been accepted as representing electrons.

• A line or lines can be used to represent a single bond or double/triple bonds between atoms.

Don’t be daft!

non-bonding/lone pairs/valence electrons you absolutely MUST add them in.

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